Emission Spectra of He₂* Excimer Produced from the He₂+/2e⁻ Collisional Radiative Recombination in the He Flowing Afterglow

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The He₂^{+/}/2e⁻ collisional radiative recombination (CRR) was studied by observing He₂* excimer emissions in the 200–990 nm region. He₂* emissions from twenty-two triplet states and twelve singlet states with excitation energies of 19.3–21.8 eV were identified. The relative formation rate constant of an upper *u* level of He₂*, $k_0(u)$, rapidly decreased with increasing the excitation energy of He₂*. Major triplet and singlet He₂* states were the lowest observed $c^3\Sigma_g^+$ and $C^1\Sigma_g^+$ states, which occupied 39.9% and 7.1% of $\sum_u k_0(u)$, respectively. The total formation ratio of triplet/singlet states was 5.1. No vibrational excitation was observed for all observed He₂* states. The rotational distributions of $f^3\Sigma_u^+(v'=0)$, $e^3\Pi_g(v'=0)$, and $c^3\Sigma_g^+(v'=0)$ were expressed by double Boltzmann rotational temperatures of 310–500 K and 580–990 K for low and high rotational levels, respectively, whereas that of $d^3\Sigma_u^+(v'=0)$ was expressed by a single Boltzmann rotational temperature of 650 K. Rovibrational distributions of the f, e, d, and C states indicated that most of the excess energies (95.2–98.5%) were deposited into relative translational energies of products. The observed vibrational and rotational distributions were lower than those of statistical prior ones calculated assuming a long lived [He₂*–e⁻] intermediate.

Key words: He₂⁺ ion, Collisional radiative recombination, Flowing afterglow, He₂^{*} excimer emission, Relative formation rate constant, Rotational temperature, Statistical prior distribution

1. Introduction

There has been continuing interest in electron-ion recombination processes in discharge plasma because they are important loss processes of charged species from natural plasmas in space to man-made plasmas such as discharge plasma.^{1,2)} Since helium is an important rare gas for the formation of stable discharge plasma, electron-ion recombination processes in helium discharge plasma have been extensively studied.³⁻⁹⁾ The main purpose of previous experimental and theoretical studies has been the determination of the responsible recombination processes and their recombination coefficient or cross section at various electron densities, He pressures, and temperatures.

When the He gas pressure is low (p < about 1Torr = 133.3 Pa), it is recognized that the threebody process He⁺ + 2e⁻ \rightarrow He^{*} + e⁻ is the main recombination mechanism if the electron density is not too low or the electron temperature too high. When the He gas pressure is higher (p > about 5 Torr) and He₂⁺ is the dominant ion, the following electron-He₂⁺ recombination processes can occur:

$$He_{2^{+}} + e^{-} \rightarrow neutral products,$$

Dissociation recombination (DR), (1)

 $\begin{array}{ll} \mathrm{He_{2^{+}}+e^{-}+e^{-}} & \rightarrow \mathrm{He_{2^{*}}+e^{-},}\\ \mathrm{Collisional\ radiative\ recombination\ (CRR),} & (2) \end{array}$

$$\begin{array}{l} \text{He}_{2}^{+} + \text{e}^{-} + \text{He} \rightarrow \text{He}_{2}^{*} + \text{e}^{-}, \\ \text{Ternary recombination.} \end{array}$$
(3)

Berlande et al.⁶⁾ determined recombination coefficients of processes (1)-(3) at various electron densities and He pressures at 300 K. In general DR process (1) is slow and CRR process (2) is a major process at low He pressure and ternary recombination (3) becomes important with increasing the He gas pressure.

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Compared with the large number of kinetic studies of the He discharge plasma, there are only a few experimental studies on determining the final electronic states of neutral He₂ molecules in the electron- He_2^+ recombination process. Two pioneering flowing-afterglow (FA) optical spectroscopic studies have been carried out on the electron-He2+ recombination process leading to He2* excimer by Schmeltekopf and Broida¹⁰⁾ and Collins and Robertson¹¹⁾ in Schmeltekopf 1963 - 1964.and Broida¹⁰⁾ several He₂* emission systems observed resulting from CRR (2). Principal He₂* emission systems were 4650 Å $e^{3}\Pi_{g}(3p\pi)-a^{3}\Sigma_{u}^{+}$ and 5733 Å $f^{3}\Delta_{u}(3d\delta)$ -b³ Π_{g} systems. They found that the rotational populations of high and low rotational levels of the $e^{3}\Pi_{g}$ state were expressed by double Boltzmann temperatures of 903 K and 492 K at pdn (downstream pressure)¹⁰⁾ = 1.4 Torr and 878 and 278 K at $p_{dn} = 1.7$ Torr, respectively. Collins and Robertson¹¹⁾ identified fourteen He₂* bands with the wavelength near the band origin at 3680, 3984, 4400, 4430, 4450, 4540, 4650, 5130, 5730, 5880, 5950, 6100, 6300, and 6400 Å. Among fourteen emissions, five systems at 3680, 3984, 4450, 4650, and 5130 Å were assigned to the $4p \ {}^{3}\Pi_{g} - 2s \ {}^{3}\Sigma_{u}^{+}$, $5d \ {}^{3}\Pi_{u} - 2p \ {}^{3}\Pi_{g}$, $4d \,{}^{3}\Sigma_{u}{}^{+}-2p \,{}^{3}\Pi_{g}, \, 3p \,{}^{3}\Pi_{g}-2s \,{}^{3}\Sigma_{u}{}^{+}, \text{ and } 3p \,{}^{1}\Pi_{g}-2s \,{}^{1}\Sigma_{u}{}^{+}$ transitions, respectively. Based on the axial variation and pressure dependence of the intensity of these He2* emissions and He2+ concentration, they identified the dominant populating mechanism of He_2^* to be CRR (2).¹¹⁾ After these pioneering studies, no detailed optical studies have been reported for CRR process (2) during the past about 60 years.

We have previously made a systematic optical spectroscopic studies on the He⁺/2e⁻, Ne⁺/2e⁻, and Ar⁺/2e⁻ CRR reactions by observing He*. Ne*. and Ar* emissions in the FA.¹²⁻¹⁵⁾ In the He⁺/2e⁻ CRR reaction, the formation of fifty-one singlet and triplet ns, np, and *nd* Rydberg states of He* in the 23.01-24.53 eV range was observed. The electronic-state distribution decreased with increasing the excitation energy of He^{*}. They were expressed by double Boltzmann electronic distributions with effective electronic temperatures of 0.46 eV in the 22.7-24.4 eV range and 0.089 eV in the 24.4–24.5 eV range.

In this study, the $\text{He}_2^+/2e^-$ CRR reaction (2) is investigated by observing ultraviolet-visiblenear infrared emissions of He_2^* excimer in the He FA. The relative formation rate constants are determined. The vibrational and rotational distributions are determined and compared with those predicted from statistical prior distributions. Preliminary results of this paper were communicated in a review article.¹⁶⁾ Although the contribution of radiative cascade from upper He₂* states to the lower levels was not considered in the determination of relative formation rate constant in the previous study,¹⁶⁾ we take it into account in this study.

2. Experimental

The FA apparatus used in this study has been previously described in detail.¹⁶⁾ In brief, the flow reactor, which consists of a quartz discharge tube (11 mm i. d.) and a stainlesssteel reaction cell (60 mm i. d.), was continuously evacuated by means of a 10,000 liter/min mechanical booster pump combined with a 1,600 liter/min oil rotary pump. $He(2^{3}S)$, He⁺, He₂⁺, electrons were generated by a microwave discharge of high purity helium gas (purity >99.995%) in a discharge flow operated at 1.6-20 Torr. The mechanical booster pump was equipped with a continuously variable gate valve.¹⁶⁾ Opening or closing the gate valve allowed us to vary the flow tube pressure from 1.8 Torr to 20 Torr. Low pressure experiments at 1.6-1.8 Torr were carried out by completely opening the gate valve, whereas high pressure experiments at 2-20 Torr were conducted by partially closing the gate valve.

The contribution of He⁺ and He₂⁺ ions to the observed emissions was examined using an ioncollector grid placed between the discharge section and the reaction zone. The efficiency of ion collection was found to be >98% by observing the N₂⁺(C²Σ_u⁺-X²Σ_g⁺) emission resulting from the He⁺ + N₂ charge-transfer reaction.¹⁷⁾

The positive He_{2^+} ions were formed by the three-body reaction of He^+ with 2He in a flow tube:

$$\begin{array}{ccc} & & & & \\ & & & & \\ & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & &$$

А reaction flame. observed 10 cmdownstream from the center of the discharge, was dispersed in the 200–990 nm region with a Spex 1250M monochromator equipped with photomultiplier cooled (Hamamatsu а Photonics R376 or R316-02). The wavelength response of monochromator and the optical detection system were corrected using standard D₂ and halogen lamps.

3. Results and Discussion

3.1 Emission spectra of He_2^* excimer resulting from the $He_2^+/2e^-$ CRR reaction

Figure 1 shows typical emission spectra observed in the 330-980 nm region at a He gas pressure of 1.8 Torr. Thirty triplet and twelve singlet transitions of He₂* given in Table 1 are identified by referring to reported molecular spectral data of He₂*.¹⁹⁾ The assignment of several weak transitions of He₂* is not shown in Fig. 1 for the sake of clarity. In addition to many He₂* emissions, He*, H*, and N₂+(B-X) emissions are identified.^{20,21)} He* lines are dominantly formed by the He⁺/2e⁻ CRR reaction,^{12,13)} whereas H* lines and N₂⁺ emission arise from residual H₂O and N₂, respectively.^{21,22)} The observed He₂* emissions disappeared when charged species were removed from the He afterglow on applying an electrostatic potential to the grid. These results imply that the He₂* states are produced through electron-ion recombination process.

The relative intensities of the observed He_2 transitions are listed in Table 1. There are



Fig. 1. Emission spectra of He₂* resulting from the He₂+/2e⁻ CRR in the He FA at a He gas pressure of 1.8 Torr. Lines marked by * and \diamondsuit are He* and H* lines, respectively.

three ${}^{3}\Delta_{u}(nd\delta)$, ${}^{3}\Pi_{u}(nd\pi)$, and ${}^{3}\Sigma_{u}{}^{+}(nd\sigma)$ components in the q, m, j, and f triplet states with n = 6, 5, 4, and 3, respectively. Although it was difficult to determine relative intensities of

Table 1. Relative intensities of He_2^* excimer bands produced through the $\text{He}_2^+/2e^-$ CRR in the He FA.

Transition	RI ^{a)}	Transition	RI
$q^{3}\Delta_{u}(6d\delta)$ - $b^{3}\Pi_{g}$	J		
$q^3\Pi_u(6d\pi)$ - $b^3\Pi_g$	- 4.5		
$q^3\Sigma_u{}^+\!(6d\sigma){}{}^+\!b^3\Pi_g$]		
$q^{3}\Delta_{u}(6d\delta)\text{-}c^{3}\Sigma_{g}^{+}$	٦		
$q^3\Pi_u(6d\pi)\text{-}c^3\Sigma_g^{+}$	- 1.9		
$q^3\Sigma_u{}^+\!(6d\sigma){\textbf{-}}c^3\Sigma_g{}^+$]		
$p^3\Pi_g(6p\pi)\text{-}a^3\Sigma_u^{+}$	1.4	$P^1\Pi_g(6p\pi)\text{-}A^1\Sigma_u^{+}$	1.3
$o^{3}\Sigma_{u}{}^{\scriptscriptstyle +}(6s\sigma)\text{-}c^{3}\Sigma_{g}{}^{\scriptscriptstyle +}$	1.1		
$o^{3}\Sigma_{u}{}^{+}(6s\sigma)\text{-}b^{3}\Pi_{g}$	1.1		
$m^{3}\Delta_{u}(5d\delta)\text{-}c^{3}\Sigma_{g}{}^{+}$	ך		
$m^3\Pi_u(5d\pi)\text{-}c^3\Sigma_g{}^+$	- 3.0		
$m^{3}\Sigma_{u}^{+}(5d\sigma)$ - $c^{3}\Sigma_{g}^{+}$	J		
$m^{3}\Delta_{u}(5d\delta)\text{-}b^{3}\Pi_{g}$	J	$M^1\Delta_u(5d\delta)$ - $B^1\Pi_g$ ך	
$m^3\Pi_u(5d\pi)\text{-}b^3\Pi_g$	- 5.6	$M^{1}\Pi_{u}(5d\pi)$ - $B^{1}\Pi_{g}$	- 1.0
$m^3\Sigma_u{}^+(5d\sigma){}^+b^3\Pi_g$]	$M^{1}\Sigma_{u}^{+}(5d\sigma)-B^{1}\Pi_{g}$	
$l^3\Pi_g(5p\pi)\text{-}a^3\Sigma_u{}^+$	3.6	$L^1\Pi_g(5p\pi)\text{-}A^1\Sigma_u^{+}$	2.4
$k^{3}\Sigma_{u}{}^{\scriptscriptstyle +}(5s\sigma)\text{-}c^{3}\Sigma_{g}{}^{\scriptscriptstyle +}$	1.3		
$k^3\Sigma_u{}^+(5s\sigma){}^+b^3\Pi_g$	1.5		
$k'^{3}\Sigma_{g}^{+}\!(5p\sigma)\text{-}a^{3}\Sigma_{u}^{+}$	1.2		
$j^3\Delta_u(4d\delta)$ - $b^3\Pi_g$	ן		
$j^3\Pi_u(4d\pi)$ - $b^3\Pi_g$	-17.0		
$j^3\Sigma_u{}^+(4d\sigma){}^+b^3\Pi_g$]		
$i^3\Pi_g(4p\pi)\text{-}a^3\Sigma_u{}^+$	6.3		
$h^3\Sigma_u^{+}\!(4s\sigma)\!\!\cdot\!\!b^3\Pi_g$	2.2	$H^1\Sigma_u^{+}\!(4s\sigma)\text{-}B^1\Pi_g$	1.3
$f^{3}\Delta_{u}(3d\delta)\text{-}b^{3}\Pi_{g}$	14.1	$F^1\Delta_{u}(3d\delta)\text{-}B^1\Pi_{g}$	3.4
$f^3\Pi_u(3d\pi)\text{-}b^3\Pi_g$	15.0	$F^{1}\Pi_{u}(3d\pi)$ - $B^{1}\Pi_{g}$	3.1
$f^3\Sigma_u{}^+\!(3d\sigma){\text -}b^3\Pi_g$	11.9	$F^1\Sigma_u{}^+\!(3d\sigma){\textbf -}B^1\Pi_g$	1.9
$e^3\Pi_g(3p\pi)\text{-}a^3\Sigma_u{}^+$	10.5	$E^1\Pi_g(3p\pi)\text{-}A^1\Sigma_u^{+}$	1.9
$d^3\Sigma_u{}^+(3s\sigma){}\text{-}b^3\Pi_g$	13.2	$D^{1}\Sigma_{u}{}^{+}(3s\sigma)\text{-}B^{1}\Pi_{g}$	4.6
$c^{3}\Sigma_{g}^{+}(3p\sigma)$ - $a^{3}\Sigma_{u}^{+}$	100	$C^{1}\Sigma_{g}^{+}(3p\sigma)$ - $A^{1}\Sigma_{u}^{+}$	17.7

a) Relative intensity

the three components for the upper q, m, and j states because of their heavy overlapping in the narrow wavelength region. Therefore, total intensities of the three components in each system are given in Table 1. On the other hand, three components were observed separately for the f state, so that their relative intensities are given in Table 1.

Figures 2 and 3 show energy-level diagrams of triplet and singlet states of He_2^* , where observed emitting levels are shown in red. The excitation energies of He_2^* and the



Fig. 2. Energy-level diagram of triplet states of He₂*. Observed emitting levels in the He₂^{+/2e⁻} CRR reaction are shown in red.



Fig. 3. Energy-level diagram of singlet states of He_2^* . Observed emitting levels in the $He_2^{+/2e^{-}}CRR$ reaction are shown in red.

recombination energy of He₂⁺ (22.223 eV) were obtained from Ref. 19. Although all of the He₂* states below the recombination energy of He₂⁺ can be formed in the He₂⁺/2e⁻ CRR reaction, 22 triplet levels from c to q states in the 19.2–21.9 eV region and 12 singlet levels from C to P states in the 19.5–21.8 eV region were observed in this study. These results indicate that product He₂* are distributed in a wide energy range in both triplet and singlet states. A similar wide product-state distribution has been observed in the formation of He* from the He⁺/2e⁻ CRR reaction.¹³⁾

3.2 Relative formation rate constants of He_2^* excimer in the $He_2^*/2e^-$ CRR reaction

emission intensities of each Relative emission system, given in Table 1, are proportional to their emission rate constants. When there is no nonradiative decay and pumping from the observation region, the emission rate constants correspond to their formation ones. To the best of our knowledge, no nonradiative decay processes have been known for the He₂* states. Radiative lifetimes of the d, e, and f states have been measured as 25±5 ns, 57±10 ns, and 19±5 ns, respectively.²³⁾ Although radiative lifetimes of other excited states observed in this study are unknown, they may be the same order as the above states. Radiative lifetimes of excited He₂* molecules $(\leq 57 \text{ ns})$ are expected to be short enough to emit radiation within the observation region. It is therefore reasonable to assume that the relative emission rate constants reflect their formation ones.

The overall rate constant of the production of an upper state of He_2^* , k(u), is given by summation of direct formation rate constant, $k_0(u)$, and the contribution of radiative cascade from upper states, k(u: cascade),

$$k(u) = k_0(u) + k(u: \text{cascade}).$$
(5)

Actually radiative cascade partially contributes to the formation of the lowest energy c state (Table 1). Its $k_0(u)$ value was estimated by subtracting the contribution of radiative cascade. Results obtained are given in Table 2, where T_0 or T_e values of He^{2*} states are shown. When vibrational constants of He^{2*} states are known, T_0 values are calculated. On the other hand, when vibrational constants of He^{2*} are unknown, T_e values are given in Table 2. The dependence of $k_0(u)$ on the energy of He^{2*} state is shown in Fig. 4. 令和5年度

The $k_0(u)$ values decrease rapidly with increasing excitation energy of He₂* for both the triplet and singlet states. They are largest for the lowest $c^{3}\Sigma_{g}^{+}$ and $C^{1}\Sigma_{g}^{+}$ states. The total formation ratio of triplet/singlet states was 5.1, which was larger than a statistical ratio of 3.0 by 70%.

3.3 Vibrational distribution of He₂* and rotational distributions of He₂(f, e, d, c: v' = 0) in the He₂+/2e⁻ CRR reaction

Only (0,0) bands were observed for all the He₂* systems, so that no vibrational excitation was observed. Equilibrium internuclear

distances of He₂* are given in Table 2.¹⁹⁾ It is clear that equilibrium internuclear distances of the observed He₂* states (1.068–1.097 Å) are similar to that of He₂+ (X²Σ_u+:1.081 Å).¹⁹⁾ Thus, one reason for the lack of vibrational excitation is large Franck-Condon factors (FCFs) for the He₂+(X: v'' = 0) \rightarrow He₂* recombination due to similar equilibrium internuclear distance.

In order to confirm the validity of the above explanation, we calculated FCFs from the $\text{He}_2^+(X:v''=0)$ to major $\text{He}_2(d, c, \text{ and } C: v'=0-2)$ vibrational states using Morse potentials.²⁴⁾ Results obtained are shown in Table 3. The FCFs for the 0-0 transitions are

Table 2. Observed triplet and singlet excited states, energies (E_{state}), relative direct formation rate constants ($k_0(u)$), and equilibrium internuclear distance (r_e) of He₂* excimer.

State	E_{state}	$k_0(u)$	$r_e^{\rm a}$	State	E _{state}	$k_0(u)$	r _e
	(eV)		(Å)		(eV)		(Å)
$q^{3}\Delta_{u}(6d\delta)$	21.8 (T_e)	7					
$q^3\Pi_u(6d\pi)$	21.8 (T_e)	- 6.9	1.090				
$q^{3}\Sigma_{u}^{+}(6d\sigma)$	21.8 (T_e)	J					
$p^3\Pi_g(6p\pi)$	21.8	1.5	1.080	$P^{1}\Pi_{g}(6p\pi)$	21.8	1.4	1.080
$o^{3}\Sigma_{u}^{+}(6s\sigma)$	21.8	2.4	1.089				
$m^{3}\Delta_{u}(5d\delta)$	21.7	٦		$M^1\Delta_u(5d\delta)$	21.7	ſ	
$m^3\Pi_u(5d\pi)$	21.7	-9.3	1.091	$M^{1}\Pi_{u}(5d\pi)$	21.7	- 1.0	1.091
$m^{3}\Sigma_{u}{}^{+}(5d\sigma)$	21.7			$M^{1}\Sigma_{u}{}^{+}(5d\sigma)$	21.7	J	
$l^3\Pi_g(5p\pi)$	21.7	3.9	1.080	$L^{1}\Pi_{g}(5p\pi)$	21.7	2.6	1.079
$k^{3}\Sigma_{u}{}^{\scriptscriptstyle +}\!(5s\sigma)$	21.6	3.0	1.079				
$k'^{3}\Sigma_{g}^{+}(5p\sigma)$	21.5	1.3	1.068				
$j^{3}\Delta_{u}(4d\delta)$	21.4	٦	1.081				
$j^3\Pi_u(4d\pi)$	21.4	-6.1	1.083				
$j^3\Sigma_u{}^+\!(4d\sigma)$	21.4						
$i^3\Pi_g(4p\pi)$	21.3	6.8	1.079				
$h^3\Sigma_u{}^+\!(4s\sigma)$	21.3	2.4	1.077	$\mathrm{H}^{1}\Sigma_{u}{}^{+}(4s\sigma)$	21.3	1.4	1.077
$f^{3}\Delta_{u}(3d\delta)$	20.7	15.2	1.079	$F^1\Delta_u(3d\delta)$	20.7	3.7	1.079
$f^3\Pi_u(3d\pi)$	20.7	16.2	1.087	$F^{1}\Pi_{u}(3d\pi)$	20.7	3.3	1.085
$f^{3}\Sigma_{u}{}^{+}\!(3d\sigma)$	20.6	12.8	1.091	$F^1\Sigma_u{}^+\!(3d\sigma)$	20.7	2.1	1.089
$e^{3}\Pi_{g}(3p\pi)$	20.6	11.4	1.075	$E^{1}\Pi_{g}(3p\pi)$	20.7	2.0	1.076
$d^{3}\Sigma_{u}{}^{+}\!(3s\sigma)$	20.5	14.2	1.071	$D^1\Sigma_u^{+}\!(3s\sigma)$	20.6	5.0	1.069
$c^{3}\Sigma_{g}^{+}(3p\sigma)$	19.3	100	1.097	$C^{1}\Sigma_{g}^{+}(3p\sigma)$	19.6	19.1	1.093

a) Ref. 19



Fig. 4. Relative formation rate constants of triplet and singlet states of He_2^* produced from the $He_2^{+/2}e^{-}$ CRR in the He FA.

Table 3. Franck-Condon factors between $\text{He}_2^+(X^2\Sigma_u^+: v'' = 0)$ and $\text{He}_2(d,c,C: v' = 0-2)$ states.

v'' - v'	0 - 0	0 - 1	0 - 2
$He_2^+ \rightarrow d^3 \Sigma_u^+$	0.995	$0.481(-2)^{a}$	0.279(-4)
$\mathrm{He_2}^+ \rightarrow c^3 \Sigma_g^+$	0.992	0.771(-2)	0.165(-3)
$He_2^+ \rightarrow C^1 \Sigma_g^+$	0.992	0.771(-2)	0.165(-3)

larger than 0.99 in all the three systems. These data support why vibrational excitation is insignificant for the formation of He_2^* in the $He_2^+/2e^-$ CRR reaction.

Rotational populations of He₂* were estimated for four typical band systems, for which rotational structures are well resolved. Figures 5a-5d show expanded emission spectra of $_{\mathrm{the}}$ $f^{3}\Sigma_{u}$ + $-b^{3}\Pi_{g}$, $e^{3}\Pi_{g}-a^{3}\Sigma_{u}^{+}$ $d^{3}\Sigma_{u}$ ⁺ $-b^{3}\Pi_{g}$, and $c^{3}\Sigma_{g}$ ⁺ $-a^{3}\Sigma_{u}$ ⁺ systems of He₂, where (0,0) bands of the four systems are observed. In ⁴He, the nuclear spin is zero and due to the symmetry rules for homonuclear molecules all even or odd numbered rotational levels are missing.²⁵⁾ Therefore, only odd or even levels of P, Q, or R branches are observed in the above four transitions.²⁶⁻²⁸⁾

Rotational distributions of the f, e, d, and c states were determined from relative intensities of Q or P branch. Rotational levels of He₂* which belong to Hund's case (b) are represented by the quantum number N'. The band intensity (photons s⁻¹) of a transition from a (v' N') level to a (v''N'') level is expressed as



Fig. 5. Rotational structures of the (0,0) bands of He₂(f-b, e-a, d-b, c-a) systems resulting from the He₂+/2e⁻ CRR in the He FA at a He gas pressure of 1.8 Torr.

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$$I_{v'N'v''N''} \propto N_{v'N'} Re^2 (\bar{r}_{v'v''}) q_{v'v''} v_{v'N'v''N''}^3 S_{N'N''} / g_{N'},$$
(6)

where $N_{v'N'}$ is the rotational population in a given vibrational level, $Re(\bar{r}_{v'v''})$ the electronic transition moment, $q_{v'v''}$ the FCF, $v_{v'N'v''N''}$ the transition frequency, $S_{N'N''}$ the rotational line strength, and $g_{N'} = 2N' + 1.2^{50}$ When a fixed (v', v'') band, $Re(\bar{r}_{v'v''})$ and $q_{v'v''}$ values are constant. Assuming a Boltzmann distribution, Boltzmann rotational temperature is obtained from the following relation,

$$\ln\left(\frac{I_{N'N''}}{v_{N'N''}^{3}S_{N'N''}}\right) \propto const - hcB_{\nu}N'(N'+1)/kT_{Rot}, \quad (7)$$

where k and T_{Rot} are the Boltzmann constant and the Boltzmann rotational temperature, respectively.25) In Figs. 6a-6d. $\ln(I_{N'N''}/v_{N'N''}^3S_{N'N''})$ values are plotted against N'(N'+1) for the v'=0 levels of the f, e, d, and c and states. An approximately linear relationship, i.e., approximately a Boltzmann distribution, is obtained for the above four levels. From the slopes, $hcB_{\nu\prime}/kT_{Rot}$, the effective rotational temperatures, T_{Rot} , are estimated to be 580 ± 20 K for N'=5-13 of He₂(f), 990 \pm 190 K for N'= 10-16 of He₂(e), and 830 \pm 20 K for N' = 7 - 17 of He₂(d), respectively. However, the populations at lower N' levels appear to deviate from the linear relationship. Their effective rotational temperatures are 310±70 K for N'=1-5 of He₂(f), 500±20 K for N'=2-10 of He₂(e), and 440 \pm 50 K for *N*'=1-7 of He₂(d). The rotational population of He₂(c) is expressed by a single Boltzmann distribution with an effective rotational temperature of 720±140 K. Above results are summarized in Table 4. The rotational temperatures of He₂(e) for low and high N' levels are in good agreement with

Table 4. Rotational temperatures of $He_2(f,e,d,c)$ produced from the $He_2^+/2e^-$ CRR reaction.

Emitting species	<i>T_r</i> / K	
$He_2(f^3\Sigma_u^+: v'=0)$	580±20 (N'=5-13) 310±70 (N'=1-5)	
He ₂ ($e^{3}\Pi_{g}: v'=0$)	990±190 (N'=10-16) 500±20 (N'=2-10)	
He ₂ ($d^{3}\Sigma_{u}^{+}: v'=0$)	830±20 (N'=7-17) 440±50 (N'=1-7)	
He ₂ ($c^{3}\Sigma_{g}^{+}: v'=0$)	720±140 (N'=0–12)	



Fig. 6. Estimation of effective rotational temperatures from the slopes of $\ln(I_{N'N''}/v_{N'N''}^3S_{N'N''})$ against N'(N'+1) for the He₂(f:v'=0), He₂(e:v'=0), He₂(d:v'=0), and He₂(c:v'=0) states in the He₂⁺/2e⁻ CRR reaction.

previous reported values of 492 and 903 K at $p_{dn} = 1.4 \text{ Torr.}^{10}$

In order to examine collisional relaxation, dependence of rotational populations on the buffer He gas pressure was examined in the 1.8-20 Torr range. The rotational distributions of He₂ (f: v' = 0) and He₂ (d: v' = 0) were essentially independent of the He pressure in the 1.8-20 Torr range, indicating that collisional relaxation within short radiative lifetimes of $f(19\pm5 \text{ ns})$ and $d(25\pm5 \text{ ns})$ states²³⁾ was insignificant under our experimental conditions. The rotational distributions of $\text{He}_2(f; v' = 0), \text{He}_2(e; v' = 0), \text{ and } \text{He}_2(d; v' = 0)$ reproduced by double Boltzmann rotational temperatures. Similar double Boltzmann rotational distributions have been found for the state.¹⁰⁾ Therefore, double Boltzmann е rotational distributions may be general features for the formation of He_2^* in the He₂⁺/2e⁻ CRR reaction.

From the observed rovibrational distributions of $He_2(f,e,d,c)$, we estimated the average vibrational and rotational energies of each state, denoted as $\langle E_v \rangle$ and $\langle E_r \rangle$, and the average yields of total available energy into these degrees of freedom, denoted as $\langle f_{v} \rangle$ and $< f_r >$, respectively. Here, the total available energy, E_{tot} , was estimated from the relation at 300 K: $E_{tot} = -\Delta H_0^{\circ} + 5/2RT$. From the $\langle E_v \rangle$, $\langle E_r \rangle$, $\langle f_v \rangle$ and $\langle f_r \rangle$ values, the relative translational energy $\langle E_t \rangle$ and the average yield of total available energy into the relative translational energy $\langle f_t \rangle$ were determined. Results obtained are summarized in Table 5. The $< f_r >$ value decreases with increasing the $E_{\rm tot}$ value. This implies that the total excess energy is not efficiently converted into the rotational energy of He₂*. On the basis of energetic data given in Table 5, most of total available energies (95.2-98.5%) are released to relative translational energies of products. This implies that a third-body electron efficiently receives an excess energy as its translational energy during neutralization reaction.

3.4 Comparison between observed and statistical prior rotational and vibrational distributions of He₂* excimer

In order to obtain information on reaction dynamics, statistical prior rotational and vibrational distributions,²⁹⁾ were calculated assuming a long lived $[He_2^*-e^-]$ complex:

Table 5. Average vibrational, rotational, and translational energies (eV) deposited into $He_2(f,e,d,c)$, total available energy (E_{tot} : eV), and average fractions of vibrational, rotational, and translational energies (%) deposited into $He_2(f,e,d,c) + e^-$ in the $He_2^+/2e^-$ CRR reaction.

	He ₂ (f)	$He_2(e)$	He ₂ (d)	He ₂ (c)
$< E_v >$	0.00	0.00	0.00	0.00
$< E_r >$	0.079	0.056	0.059	0.043
$E_{\rm tot}$	1.65	1.65	1.79	2.97
$< f_v >$	0.00	0.00	0.00	0.00
$< f_r >$	4.8	3.4	3.3	1.5
$< f_v > + < f_r >$	4.8	3.4	3.3	1.5
$< f_t >$	95.2	96.6	96.7	98.5

$$\operatorname{He}_{2^{+}} + 2e^{-} \rightarrow [\operatorname{He}_{2^{*}} - e^{-}] \rightarrow \operatorname{He}_{2^{*}} + e^{-}.$$
(8)

Prior rotational and vibrational distributions were calculated from the Eqs. (9) and (10), respectively:

$$P^{o}(N') \propto (2N'+1)(E_{\text{tot}} - E_{N'})^{1/2},$$
 (9)

$$P^{o}(v') \propto (E_{\text{tot}} - E_{v'})^{3/2}.$$
 (10)

The prior rotational distributions of He₂(f), He₂(e), He₂(d), and He₂(c) are compared with observed ones in Figs. 7a–7d, respectively. The prior vibrational distributions of v'=0-5 levels of He₂(f, e, d, c) states were calculated as follow:

 $\begin{array}{l} \mathrm{He_2(f:}\ \nu'{=}0{-}5) = 1.00{:}0.83{:}0.68{:}0.55{:}0.43{:}0.34,\\ \mathrm{He_2(e:}\ \nu'{=}0{-}5) = 1.00{:}0.82{:}0.66{:}0.52{:}0.39{:}0.28,\\ \mathrm{He_2(d:}\ \nu'{=}0{-}5) = 1.00{:}0.84{:}0.69{:}0.56{:}0.44{:}0.34,\\ \mathrm{He_2(c:}\ \nu'{=}0{-}5) = 1.00{:}0.91{:}0.83{:}0.75{:}0.69{:}0.63. \end{array}$

The prior rotational and vibrational distributions predict much higher rotational and vibrational excitation than the observed ones. On the basis of these facts, He₂* molecules are not formed through long-lived $[He_2-e^-]^*$, where the excess energies are deposited into all degrees of freedom statistically.

The deviation from prior rotational distribution can often be expressed in the form of surprisal,²⁹⁾

$$I_{R} = -\ln[P(v', N')/P^{0}(v', N')] = \theta_{0} + \theta_{R}g_{N'}, (11)$$

where $g_{N'} = E_{N'}/E_{tot}$.



Fig. 7. Observed and statistical prior rotational distributions of the He₂(f: v' = 0), He₂(e: v' = 0), He₂(d: v' = 0), and He₂(c: v' = 0) states in the He₂⁺/2e⁻ CRR reaction

Figures 8a–8d show rotational surprisal plots of He₂(f: v' = 0), He₂(e: v' = 0), He₂(d: v' = 0), and He₂(c: v' = 0), respectively. When the twobody dissociation model (8) is used for $P^o(N')$, good linear relationships are found in the four surprisal plots. These results imply that an exponential gap behavior is present between the experimental and statistical distributions. From slopes of each surprisal plot, large positive surprisal parameters of 23.4, 25.6, 27.2, and 45.8 are obtained for He₂ (f: v' = 0), He₂(e: v' = 0), He₂(d: v' = 0), and He₂(c: v' = 0), respectively.

4. Summary and Conclusion

He₂* excimer emissions from twenty-two triplet states and twelve singlet states with excitation energies of 19.3-21.8 eV and 19.6 - 21.8eV region, respectively, were observed by the He₂+/2e⁻ CRR reaction in the He afterglow. It was found that more He_2^* states in wide energy range are formed in the He₂+/2e⁻ CRR reaction than those observed by Schmeltekopf and Broida10) and Collins and Robertson.¹¹⁾ The relative formation rate constant of He₂*, $k_0(u)$, rapidly decreased with increasing the excitation energy. A similar tendency was observed for He* in the He+/2e-CRR reaction,¹³⁾ indicating that CRR reaction dynamics is similar between atomic He⁺ and molecular He₂⁺ ions. Major triplet and singlet product He₂* states were the lowest observed $c^{3}\Sigma_{g}^{+}$ and $C^{1}\Sigma_{g}^{+}$ states, which occupied 39.9% and 7.1% of $\sum_{u} k_0(u)$, respectively. The total formation ratio of triplet/singlet states was 5.1. No vibrational excitation was observed for all thirty-four states. The rotational distributions of $f^{3}\Sigma_{u}^{+}(\nu'=0)$, $e^{3}\Pi_{g}(\nu'=0)$, and $d^{3}\Sigma_{u}^{+}(\nu'=0)$ were expressed by double Boltzmann rotational temperatures, whereas that of $c^{3}\Sigma_{g}^{+}(\nu'=0)$ was represented by а single Boltzmann On the basis of observed temperature. rovibrational distributions, most of the total excess energy (95.2-98.5%) is deposited into the relative translational energy of products. The observed vibrational and rotational distributions were lower than those of statistical prior ones calculated assuming a long lived $[He_2^*-e^-]$ intermediate. It was therefore concluded that He_2^* molecules are formed via long lived [He₂*–e⁻] not intermediate, where excess energies are statistically distributed to all degrees of freedom.



Fig. 8. Rotational surprisal plots for the $\text{He}_2(\text{f:} v'=0)$, $\text{He}_2(\text{e:} v'=0)$, $\text{He}_2(\text{d:} v'=0)$, and $\text{He}_2(\text{c:} v'=0)$ states in the $\text{He}_2^+/2\text{e}^-$ CRR reaction.

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